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Modern Chemistry Chapter 7 Test

Modern Chemistry Chapter 7 test. STUDY. PLAY. The basis of an ionic bond is the__ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__ noble gas. Ions that are formed from more than one atom are called__

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Modern Chemistry Chapter 7. monatomic ion. binary compounds. nomenclature. oxyanions. an ion formed from a single atom. (remember to change nonmetal.... compounds composed of two different elements; (compounds of 2.... naming system. polyatomic ions that contain oxygen.

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Modern Chemistry Chapter 7. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. dvann123. Chemical Names & Formulas, Oxidation Numbers, Using Chemical Formulas, & Determining Chemical Formulas Terms. Terms in this set (32) monatomic ion. an ion formed from a single atom. (remember to change nonmetals to -ide ending)

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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided Modern chemistry chapter 7 test answer key. 37. lithium, atomic number 3 38. carbon, atomic number 6 39 Modern chemistry chapter 7 test answer key. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and ...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical Compounds

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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